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Oxidation-reduction (redox) reactions are a classification of chemical changes that involve the transfer of electrons. An example of a redox reaction is shown in Eqn. 1, when

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magnesium metal reacts with chlorine gas.

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the processes of oxidation and reduction.

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Oxidation-reduction reactions form a major class of chemical reactions. From the reactions of oxygen with sugars, fats, and proteins that provide energy for life to the corrosion of metals, many important reactions involve the processes of oxidation and reduction.

FLI SCIENTIFIC IC.

In this lab I learned a lot about oxidation-reduction reactions, half reactions, equivalence point, and titration in general. I also got more practice on stoichiometry and writing balanced net ionic equations. This relates to what we're doing in class because many of the objectives were in this lab.

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A similar principle applies to oxidation-reduction reactions. If a solution contains a substance that can be oxidized, then the concentration of that substance can be analyzed by titrating it with...

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The colorful reactions provide great practice for students to identify changes in oxidation states, write oxidation and reduction half-reactions and balance redox equations. Includes reproducible student handouts, detailed background information, complete Teacher Notes with sample data and answers to all questions and all necessary chemicals ...

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