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Compound name Aluminum sulfide Aluminum sulfite Formula PbCl2 (NH4)PO4 Hydroiodic 54 acid SECTION 7- REVIEW MODERN CHEMISTRY HRW material copyrighted under notice appearing earlier in this work. Name Modern _ Date Chemistry • Class CHAPTER 7 HOMEWORK 7-1 (pp. 203-206) VOCABULARY Write true or false for each statement. 1.

Modern Chemistry • CHAPTER 7
CHAPTER 7 REVIEW Chemical Formulas and Chemical Compounds SECTION 1 SHORT ANSWER Answer the following questions in the space provided. 1. c In a Stock system name such as iron (III) sulfate, the Roman numeral tells us (a) how many atoms of Fe are in one formula unit. (b) how many sulfate ions can be attached to the iron atom.

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7 Chemical Formulas and Chemical Compounds
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Chromium is 6quot;4 and each oxygen is #2. 56 SECTION 7-2 REVIEW MODERN CHEMISTRY HRW material copyrighted under notice appearing earlier in this work. Name Date Class CHAPTER 7 REVIEW Chemical Formulas and Chemical Compounds SECTION 7-3 SHORT ANSWER Answer the following questions in the space provided. 1.

Chemical Formulas and Chemical Compounds
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CHAPTER 7 REVIEW Chemical Formulas and Chemical Compounds MIXED REVIEW SHORT ANSWER Answer the following questions in the space provided. 1. Write formulas for the following compounds: CuCO 3 a. copper(II) carbonate Na 2SO 3 b. sodium sulfite (NH 4) 3PO 4 c. ammonium phosphate SnS 2 d. tin(IV) sulfide HNO 2 e. nitrous acid 2.

7 Chemical Formulas and Chemical Compounds
CHAPTER 7 REVIEW Chemical Formulas and Chemical Compounds SECTION 1 SHORT ANSWER Answer the following questions in the space provided. 1. c In a Stock system name such as iron (III) sulfate, the Roman numeral tells us (a) how many atoms of Fe are in one formula unit. (b) how many sulfate ions can be attached to the iron atom.

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Modern Chemistry Chapter 7 Review Answers Section 4
The Arrangement of Electrons in Atoms chapter of this Holt McDougal Modern Chemistry Companion Course helps students learn the essential lessons associated with how electrons are arranged in atoms.

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